



DEGREE CURRICULUM

EXPERIMENTATION IN CHEMICAL AND CHEMICAL ENGINEERING I

Coordination: CUADROS DOMENECH, ROSA

Academic year 2022-23

Subject's general information

Subject name	EXPERIMENTATION IN CHEMICAL AND CHEMICAL ENGINEERING I			
Code	102342			
Semester	1st Q(SEMESTER) CONTINUED EVALUATION			
Typology	Degree	Course	Character	Modality
	Master's Degree in Leather Engineering	1	COMPLEMENTARY TRAINING	Blended learning
	Master's Degree in Leather Engineering		COMPLEMENTARY TRAINING	Attendance-based
	Not informed	3	COMPULSORY	Attendance-based
Course number of credits (ECTS)	6			
Type of activity, credits, and groups	Activity type	PRALAB		TEORIA
	Number of credits	5.5		0.5
	Number of groups	2		1
Coordination	CUADROS DOMENECH, ROSA			
Department	COMPUTER SCIENCE AND INDUSTRIAL ENGINEERING			
Teaching load distribution between lectures and independent student work	Laboratory activities 60 hours Independent study work 90 hours			
Important information on data processing	Consult this link for more information.			
Language	Catalan			
Distribution of credits	0,5 THEORY 5,5 PRALAB			

Teaching staff	E-mail addresses	Credits taught by teacher	Office and hour of attention
CANTERO GOMEZ, MARIA ROSA	rosa.cantero@udl.cat	5,7	
CUADROS DOMENECH, ROSA	rosa.cuadros@udl.cat	5,8	

Subject's extra information

It is **COMPULSORY** that the students bring the following elements of individual protection (EPI) to the practices at the laboratory.

- Laboratory gown from UdL
- Protection glasses
- Mechanical protection gloves

They can be purchased through the shop Údels of the UdL:

C/ Jaume II, 67 baixos
Centre the Cultures i Cooperació Transfronterera

<http://www.publicacions.udl.cat/>

There will be a specific service for the *Campus Universitari d'Igualada*.

The use of other elements of protection (for example caps, masks, gloves of chemical or electrical risk, etc.) will depend on the type of practice to be done. In that case, the teacher will inform of the necessity of specific EPI.

Not bringing the EPI's described or not fulfilling the norms of general security that are detailed below imply that the student can not access to the laboratories or have to go out of them. The no realisation of the practices for this reason imply the **consequences in the evaluation** of the subject that are described in this course guide.

GENERAL NORMS OF SECURITY IN LABORATORY PRACTICES

- Keep the place of realisation of the practices clean and tidy. The table of work has to be free from backpacks, folders, coats...
- No short trousers or short skirts are allowed in the laboratory.
- Closed and covered footwear is compulsory in the laboratory.
- Long hair needs to be tied.
- Keep the laboratoy gown laced in order to be protected from spills of chemicals.
- Bangles, pendants or wide sleeves are not allowed as they can be trapped.
- Avoid the use of contact lenses, since the effect of the chemical products is much bigger if they enter between the contact lense and the cornea. Protection over-glasses can be purchased.
- No food or drink is allowed in the laboratory.
- It is forbidden to smoke in the laboratories.
- Wash your hands whenever you have contact with a chemical product and before going out of the laboratory.
- Follow the instructions of the teacher and of the laboratory technicians and ask for any doubt on security.

For further information, you can check the following document of the *Servei de Prevenció de Riscos Laborals de la UdL*:
<http://www.sprl.udl.cat/alumnes/index.html>

Learning objectives

When finishing the subject the student must be able to:

- Consolidate a laboratory methodology initiated in chemistry and experimentation in chemistry subjects.
- Apply to theoretical concepts developed in other subjects of the studies.
- Know the use of the material and the devices that are in a chemical laboratory.
- Know the handling of products and safety in the chemical laboratory.
- Know how to perform in the chemical laboratory.
- Work as a team performing a proper job distribution and solving the possible conflicts that arise during its execution.

Competences

Basic

B01 That students have demonstrated to possess and understand knowledge in an area of study that starts from the base of general secondary education, and is usually found at a level that, although supported by advanced textbooks, also includes some aspects that imply knowledge coming from the vanguard of his/her field of study.

B02 That students know how to apply their knowledge to their work or vocation in a professional manner and possess the skills that are usually demonstrated through the elaboration and defense of arguments and the resolution of problems within their area of study.

B03 That students have the ability to gather and interpret relevant data (usually within their area of study) to make judgments that include a reflection on relevant social, scientific or ethical issues.

B04 That students can transmit information, ideas, problems and solutions to a specialized and non-specialized public.

B05 That students have developed those learning skills necessary to undertake further studies with a high degree of autonomy.

General competences

CG3. To synthesize basic and technological subjects, which enable them to learn new methods and theories, and provide them with versatility to adapt to new situations.

CG4. To solve problems with initiative, make decisions, creativity, critical reasoning and to communicate and transmit knowledge, skills and abilities in the field of Industrial Chemical Engineering.

CG6. To implement specifications, regulations and mandatory rules.

CG7. To analyze and assess the social and environmental impact of technical solutions.

Specific competences

CE21. To design and manage applied experimentation procedures, especially for the determination of thermodynamic and transport properties, and modeling of phenomena and systems in the field of chemical engineering, systems with fluid flow, heat transfer, material transfer operations, kinetics of chemical reactions and reactors.

Transversal

CT5. To apply essential notions of scientific thinking.

Subject contents

- Volumetric methods of analysis.
- Thermochemistry.
- Basic operations.
- Organic synthesis.

Methodology

- Attendance at practice sessions is compulsory as it involves hours of laboratory practices.
- The rest of the hours are of individual work, both for the preparation of the practices and for the subsequent realization of calculations, questions and the deduction of the conclusions deduction.
- Practices will be carried out in small groups.
- After completing the practice in the laboratory, the questions and all the calculations, which are at the end of each procedure of the practice dossier, will be presented.
- At the end of the laboratory sessions, a random practice will be chosen for a report, this report will contribute to the student's final grade.
- Each student will have to bring a laboratory notebook where the data, calculations and questions will be written down, this book will also be valued to contribute to the final grade.
- Once the laboratory sessions have finished, the student will be taking an exam, which will contribute to the final grade of the subject.

Development plan

PraLab	Week 1	Week 2	Week 3	Week 4	Week 5	Week 6	Week 7
G 1	Presentation Mohr Titration	Calibration line Aspirin	Simple distillation	Simple distillation	Rectification	Chemical kinetics	Chemical kinetics
G 2	Presentation Mohr Titration	Redox Calibration line	Chemiluminescence	Chemiluminescence Aspirin	Simple distillation	Simple distillation	Rectification
G 3	Presentation Mohr Titration	Potentiometry	Chemical kinetics	Chemical kinetics	Chemical kinetics	Chemiluminescence	Chemiluminescence Redox
G 4	Presentation Mohr Titration	Chemiluminescence	Chemiluminescence Redox	Calibration line Aspirin	Simple distillation	Simple distillation	Potentiometry
G 5	Presentation Mohr Titration	Chemical kinetics	Chemical kinetics	Chemical kinetics	Calibration line Aspirin	Simple distillation	Simple distillation

PraLab	Week 8	Week 9	Week 10	Week 11	Week 12	Week 13
G 1	Chemical kinetics	Chemiluminescence	Chemiluminescence Redox	Potentiometry	Thermochemistry	Company visit
G 2	Potentiometry	Chemical kinetics	Chemical kinetics	Chemical kinetics	Thermochemistry	Company visit
G 3	Calibration line Aspirin	Simple distillation	Simple distillation	Rectification	Thermochemistry	Company visit

G 4	Rectification	Chemical kinetics	Chemical kinetics	Chemical kinetics	Thermochemistry	Company visit
G 5	Chemiluminescence	Chemiluminescence Redox	Rectification	Potentiometry	Thermochemistry	Company visit

Evaluation

The final grade of the subject will be the sum of the following percentages:

- Questions: 20% of the average grades among all the questions presented.
- Practices report: 25% of the grade.
- Written exam: 30% of the grade.
- Lab book and laboratory work: 25% of the average mark between 50% laboratory and 50% book.

There will be a recovery test for the written exam..

Bibliography

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